-hydrofluoric acid solud 10,000 psi for 3 hrs. rent metal strength and pick-up during process-



and tungsten. X 250.

bond formed between pagation and fracture behavior is attributed sperature only increases properties of the joint. a 2200° C. in bonded

t to the molybdenumappear required to get armed at 1600° C. and the joint was relatively temperatures would be beneficial. Since only a limited Kirkendall effect was noted in heat treatments up to 2200° C., there does not appear to be an inherent problem in the use of higher bonding temperatures.

## Rhenium-to-Rhenium

Excellent bonds between rhenium surfaces can be achieved at  $1600^{\circ}$  C. and 10,000 psi for 3 hrs. A surface preparation of grinding and polishing with alumina grit followed by hydrogen cleaning at  $1100^{\circ}$  C. has been satisfactory. These bonds have strength and ductility equivalent to the base material. Fig. 12 shows a typical rhenium self-bond. Note the heavy twinning in the microstructure which resulted during preparation of the metallographic specimen.



Fig. 12. Rhenium self bond. × 250.

## Rhenium - Tungsten

Excellent bond strength can be achieved in this system; however, the bond is very brittle due to the formation of intermetallic compounds. This bond is satisfactory for compression loading applications. Fig. 13 shows a bond which was formed at  $1600^{\circ}$  C. and 10,000 psi for 3 hrs. and then heat treated at  $2200^{\circ}$  C. for 1 hr. Note that diffusion is quite sluggish and that no K i r k e n d a l l effect is evident under these conditions.

## Rhenium - Zirconium

Fig. 14 shows a bond formed between rhenium and zirconium at 1100° C. and 10,000 psi for 3 hrs. Note that excessive diffusion has occurred in this case and it would be desirable to decrease both time and temperature of the

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